138. Dendritic Styryl TADDOLs as Novel Polymer Cross-Linkers: First Application in an Enantioselective Et₂Zn Addition Mediated by a Polymer-Incorporated Titanate

Preliminary Communication

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TADDOLs ($\alpha, \alpha, \alpha', \alpha'$ -tetraaryl-1,3-dioxolane-4,5-dimethanols) with dendritic branches attached to the aryl groups and with 8 or 16 peripheral styryl double bonds have been synthesized (4 and 5). With these compounds, dendritic molecules were employed for the first time as cross-linkers in a polymerization. The resulting polystyrene was loaded with titanate (Ti(OCHMe₂)₄) to generate polymer beads incorporating Ti-TADDOLate centers (*Figs. 1* and 2) for enantioselective catalysis (Et₂Zn addition to PhCHO). Compared with conventionally polymer-attached, insoluble Ti-TADDOLates (*cf.* 3), the new materials have a much higher catalytic activity, rather close to that of soluble analogs (*Fig. 3*).

TADDOLs ($\alpha, \alpha, \alpha', \alpha'$ -tetraaryl-1,3-dioxolane-4,5-dimethanols) have been shown to be useful ligands in metal complexes for stoichiometric and catalytic enantioselective transformations²). Due to their preparation from tartrate, the original TADDOLs of C_2 -(1) and C_1 -symmetry (2) are readily modified; we have recently demonstrated that a styryl-substituted TADDOL can be incorporated in a cross-linked (divinylbenzene) polystyrene, and that the corresponding titanate 3 can, for instance, be used for highly enantioselective nucleophilic additions to aldehydes [2]. Of course, the insoluble titanate 3 was catalyzing the reactions studied less efficiently than the monomers derived from 1 or 2 (vide infra) [3]. We, therefore, decided to prepare dendritically substituted TAD-DOLs, with peripheral styryl groups, such as 4 and 5. To the best of our knowledge³) there has been no report in the literature of the use of a dendrimer as cross-linker in a polymerization. Also, we are not aware of any work (outside of our own group) in which a chiral, catalytically active site has been placed in the core of a dendrimer. It is important to emphasize that only dendrimers of low generation number can possibly serve as carriers of ligands for catalysis: the densly packed surfaces of high-generation dendrimers (with critical size) prevent catalytic activity, both, at the surface (crowding!) and

¹) Part of the projected Ph. D. thesis of P.B.R., and of the Master's thesis of H. S., ETH-Zürich, 1997.

²) For short review articles, see [1].

³) Besides a distyryl-TADDOL [2], a distyryl salen has been used as a cross-linker in a polymerization [4]. There are, of course, myriades of chiral ligands which were polymer bound or copolymerized for immobilization, see ref. in [2], and two recent articles on soluble [5] and insoluble [6] polymers containing quinine-type ligands for dihydroxylations.

in the cavities (hindered access!)⁴). We hoped that, due to the conformationally flexible dendritic spacers between the chiral ligand and the polymer backbone(s), the cross-linked, insoluble polymers obtained from compounds 4 or 5, and styrene would have readily accessible TADDOL moieties, and thus provide catalytic activity comparable to that of soluble analogs.

The synthesis of the novel cross-linkers starts with the reaction of $\{[4-(tert-buty])(imethylsilyloxy]phenyl\}$ magnesium bromide with the commercial acetonide of (R,R)-diethyl tartrate to give diol 6 (80%) which was deprotected (Bu₄NF) to yield the hexol 7 (87%) and etherified at the phenolic positions by the corresponding benzylic branch bromides to furnish⁵) 60-70% of the desired first- and second-generation dendrimers 4 and 5, respectively. The branches were obtained by *Fréchet*'s convergent methodology [8] from 4-vinylbenzyl chloride and 3,5-dihydroxybenzoate⁶). For comparison purposes (*vide infra*), we have also prepared the dendrimer 8 lacking the vinyl groups⁷) present in 4.

The styryl TADDOLs 4 and 5 were copolymerized with styrene, following the suspension procedure detailed in [2]. The size of the beads $(100-600 \,\mu\text{m})$ obtained decreased with increasing ratio styrene/styryl-TADDOL (6:1-48:1). Treatment of the thoroughly dried beads with Ti(OCHMe₂)₄ in toluene (azeotropic removal of Me₂CHOH) gave the Ti-loaded particles (accompanied by a color change, see *Fig. 1*). A schematic presentation of polymer **9a** (from **4**) is given in *Fig. 2*; **9b** (from **5**) has an analogous composition, with 16 rather than 8 attachments to the polystyrene backbone.

The new polymer-incorporated Ti-TADDOLates were tested for the catalysis of Et_2Zn addition to PhCHO under the usual conditions [2] [10]. The results, including those of a comparison with soluble Ti-TADDOLates, are shown in the *Table: i*) the dendritically cross-linked Ti-TADDOLate polymers **9** give rise to almost the same enantioselectivity as the simple polymer-bound analog **3**. *ii*) With 0.2 equiv. of Ti-TADDOLates, the enantioselectivities obtained under homogeneous (er up to 99:1) and heterogeneous (er up to 98:2) conditions are comparable. *iii*) The selectivity seems to increase with decreasing TADDOLate content in the dendritic polymers **9**. *iv*) The polymer **9b** prepared with the second-generation dendrimer **5** does not appear to give better results than the one from first-generation **4**.

In Fig. 3, we show the rates of $\text{Et}_2\text{Zn-to-PhCHO}$ additions at -25° with the different catalysts. While the polymer-bound Ti-TADDOLate 3 is considerably slower (4.5 h to 90% conversion) [3] than the soluble analogs from 2 (10 min), and from 8 (12 min), the novel dendritically incorporated Ti-TADDOLate 9a is in fact much faster (40 min) than 3. Thus, our expectation that dendritic cross-linking ligands for catalytic applications of organometallic complexes would render polymer-incorporated active sites, which are more easily accessible, was actually met!

⁴) For large dendrimers with chiral building blocks on the surface, see the papers on 'dendritic boxes' and on dendrimers as chiral catalysts – a critical note' [7].

⁵) We were surprised to observe formation of side products containing five branches (one on a tertiary C-atom of the TADDOL unit!).

⁶) See also our previous syntheses of chiral dendrimers with *Fréchet* branches [9].

⁷) In spite of the propensity for spontaneous polymerization (especially in solution), the compounds containing styryl groups, as described here, can be handled with due care; they were purified (by chromatography or recrystallization) and fully characterized (including by elemental analyses).



1 ($R^1 = R^2 = Me$) **2** ($R^1 = Ph, R^2 = H$)



3 (PS = polystyrene chain)



4 (m.p. 63-65°)



5 (m.p. 68-70°)



6 (R = TBDMS) 7 (R = H)



8 (m.p. 72-73°)



Fig. 1. Polymer beads (diameter ca. 600 μ m) formed by suspension polymerization of the dendritic styryl TADDOL 4 and styrene before (left) and after (right) titanate loading. The particles shown were obtained by fractional sieving of a bead population from styrene/17% 4.



Fig. 2. Schematic presentation of the diisopropoxy-Ti-TADDOLate **9a** derived from the cross-linked suspension copolymer obtained from **4** and styrene (ratio 1:20), after treatment with $Ti(OCHMe_2)_4$ (with removal of 2 equiv. of HOCHMe₂)

Multiple use of the new materials and applications in other reactions, together with all experimental details, will be described in a forthcoming full paper.

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Table. Enantioselective Formation of (S)-1-Phenylpropan-1-ol by Ti-TADDOLate-Catalyzed Et₂Zn Addition to PhCHO. The percentages given in square brackets for the polymer-bound Ti-TADDOLates are the mol-% amounts of the dendritic TADDOLs used in the copolymerization with styrene.

| | [∼] H + Et₂Zn | x equiv. Ti-TADDOLate 5x equiv. Ti(OCHMe ₂) ₄ toluene -25° | | |
|----------------------------------|------------------------|--|------------------------------|-------------------------------------|
| Ti-TADDOLate (x equiv.) | PhCHO [mmol] | Et ₂ Zn [mmol] | Conversion after 20 h [%] | Enantiomer ratio (er) of product |
| Homogeneous | | | | |
| from 2 (0.2) | 11 | 20 | quant. | 99:1 |
| from 8 (0.2) | 4 | 7.2 | quant. | 97:3 |
| Heterogeneous | | | | |
| $3 (0.2)^{a}$ | 11 | 20 | quant. | 98:2 |
| 9a [17% 4] (0.02) | 10 | 18 | 68 | 59:41 |
| 9b [13% 5] (0.02) | 10 | 18 | 53 | 51:49 |
| 9a [6% 4] (0.05) | 4 | 7.2 | 41 | 76:24 |
| 9a [5% 4] (0.05) | 4 | 7.2 | 53 | 83:17 |
| 9b [3% 5] (0.05) | 10 | 18 | 23 | 76:24 |
| 9a [5% 4] (0.20) | 4 | 7.2 | quant. | 94:6 |
| 9b [2% 5] (0.20) | 3 | 5.4 | quant. | 93:7 |

^a) Taken from [2].



Fig. 3. Reaction rates of the addition of Et_2Zn to PhCHO. Catalysis by the soluble Ti-TADDOLates from 2 (black) and from the dendritic compound 8 (green). Catalysis by the polymer-bound non-dendritic Ti-TADDOLate 3 (blue) and 9a (red). Ti-TADDOLate/Ti(OCHMe₂)₄ ratio is in all cases 1:5. Enantioselectivities with Ti-TADDOLates from 2 (er 99:1), 8 (er 97:3) with 3 (er 98:2), and with 9a (er 94:6).

REFERENCES

- R. Dahinden, A. K. Beck, D. Seebach, 'Encyclopedia of Reagents for Organic Synthesis', Ed. L. Paquette, J. Wiley & Sons, Chichester, 1995, Vol. 3, p. 2167; D. Seebach, A. K. Beck, Chimia 1997, 51, 293.
- [2] D. Seebach, R. E. Marti, T. Hintermann, Helv. Chim. Acta 1996, 79, 1710.
- [3] A. K. Beck, P. J. Comina, D. Seebach, hitherto unpublished results, ETH-Zürich, 1996/97.
- [4] F. Minutolo, D. Pini, P. Salvadori, Tetrahedron Lett. 1996, 37, 3375.
- [5] H. Han, K. D. Janda, J. Am. Chem. Soc. 1996, 118, 7632.
- [6] P. Salvadori, D. Pini, A. Petri, J. Am. Chem. Soc. 1997, 119, 6929.
- [7] J. F. G. A. Jansen, H. W. I. Peerlings, E. M. M. de Brabander-van den Berg, E. W. Meijer, Angew. Chem. 1995, 107, 1321; ibid., Int. Ed. Engl. 1995, 34, 1206; M. S. T. H. Sanders-Hovens, J. F. G. A. Jansen, J. A. J. M. Vekemans, E. W. Meijer, Polym. Mater. Sci. Eng. 1995, 75, 338.
- [8] C. J. Hawker, J. M. J. Fréchet, J. Am. Chem. Soc. 1990, 112, 7638.
- [9] D. Seebach, J.-M. Lapierre, G. Greiveldinger, K. Skobridis, Helv. Chim. Acta 1994, 77, 1673.
- [10] B. Schmidt, D. Seebach, Angew. Chem. 1991, 103, 1383; ibid., Int. Ed. Engl. 1991, 30, 1321; D. Seebach, D. A. Plattner, A. K. Beck, Y. M. Wang, D. Hunziker, W. Petter, Helv. Chim. Acta 1992, 75, 2171; D. Seebach, A. K. Beck, B. Schmidt, Y. M. Wang, Tetrahedron 1994, 50, 4363; Y. N. Ito, X. Ariza, A. K. Beck, A. Bohác, C. Ganter, R. E. Gawley, F. N. M. Kühnle, J. Tuleja, Y. M. Wang, D. Seebach, Helv. Chim. Acta 1994, 77, 2071.